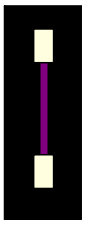
The Four Problems...

1. Atomic structure: why aren't the electrons in the nucleus?



2. Why do gases glow when electrified?

3. Why are only certain colors seen?

4. $1/\lambda = R(1/4 - 1/n^2)$?

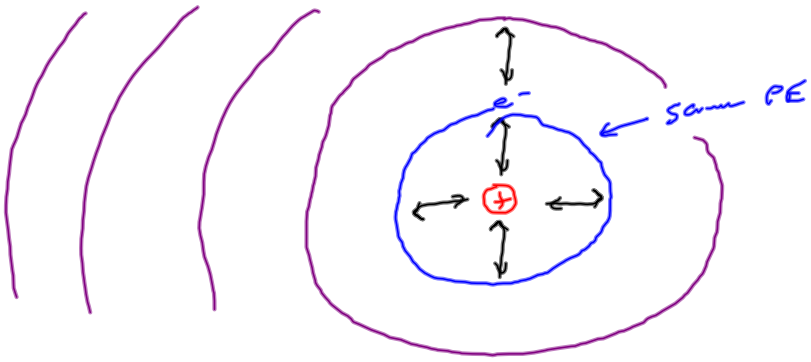
$1/\lambda = R(1/9 - 1/n^2)$?

$1/\lambda = R(1 - 1/n^2)$?

Sep 21-7:03 AM

Bohr's Answer → NRG is quantized

e⁻ NRG = Potential NRG



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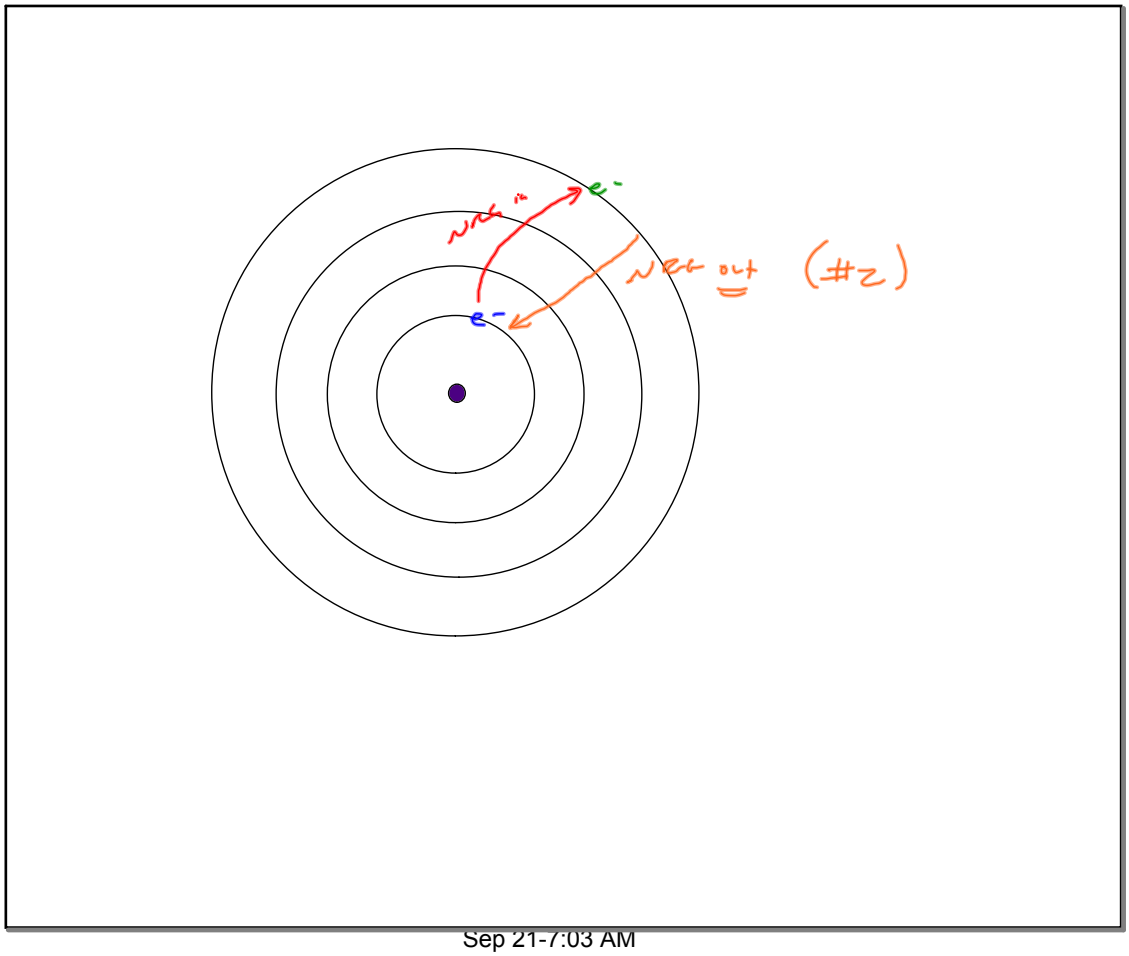
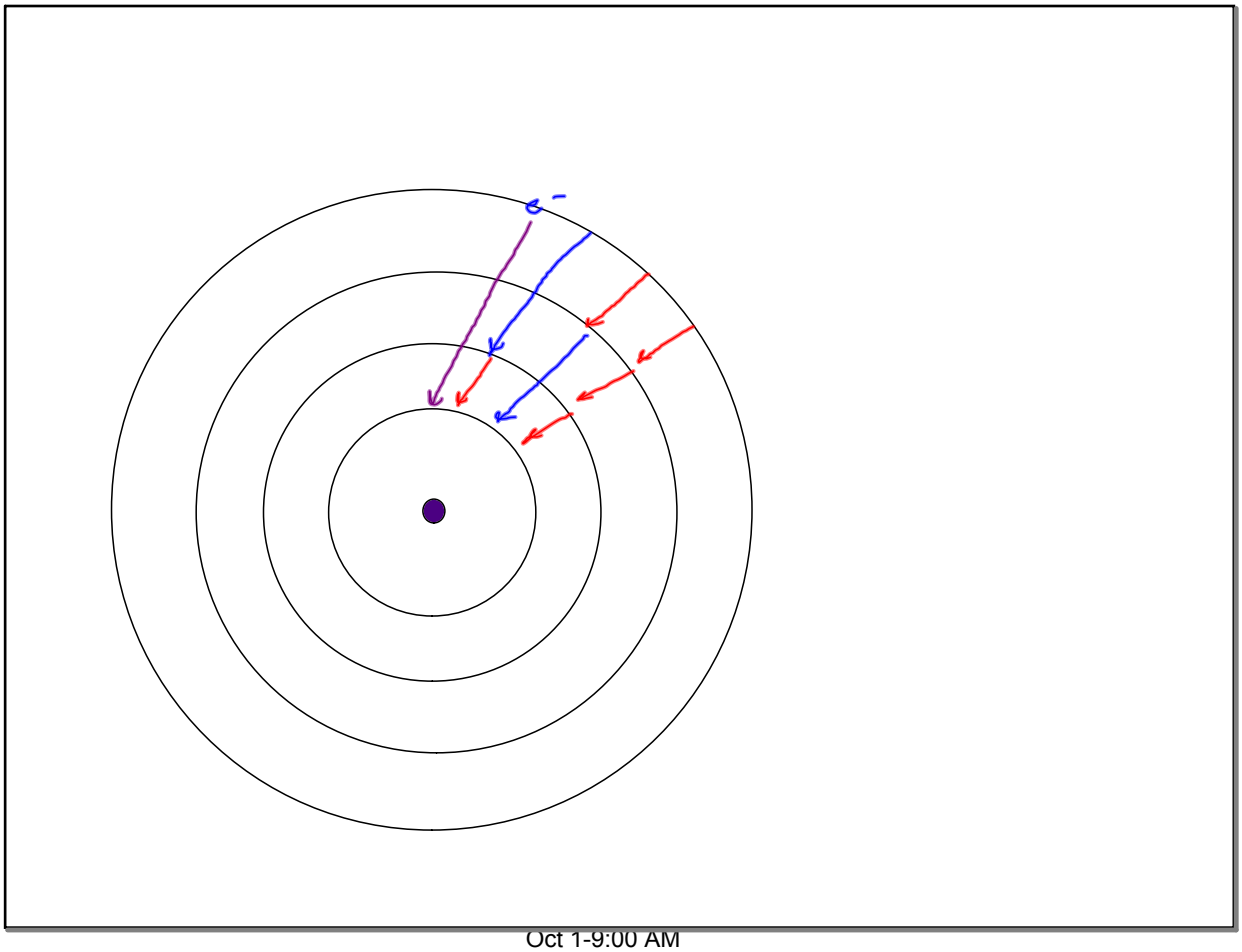
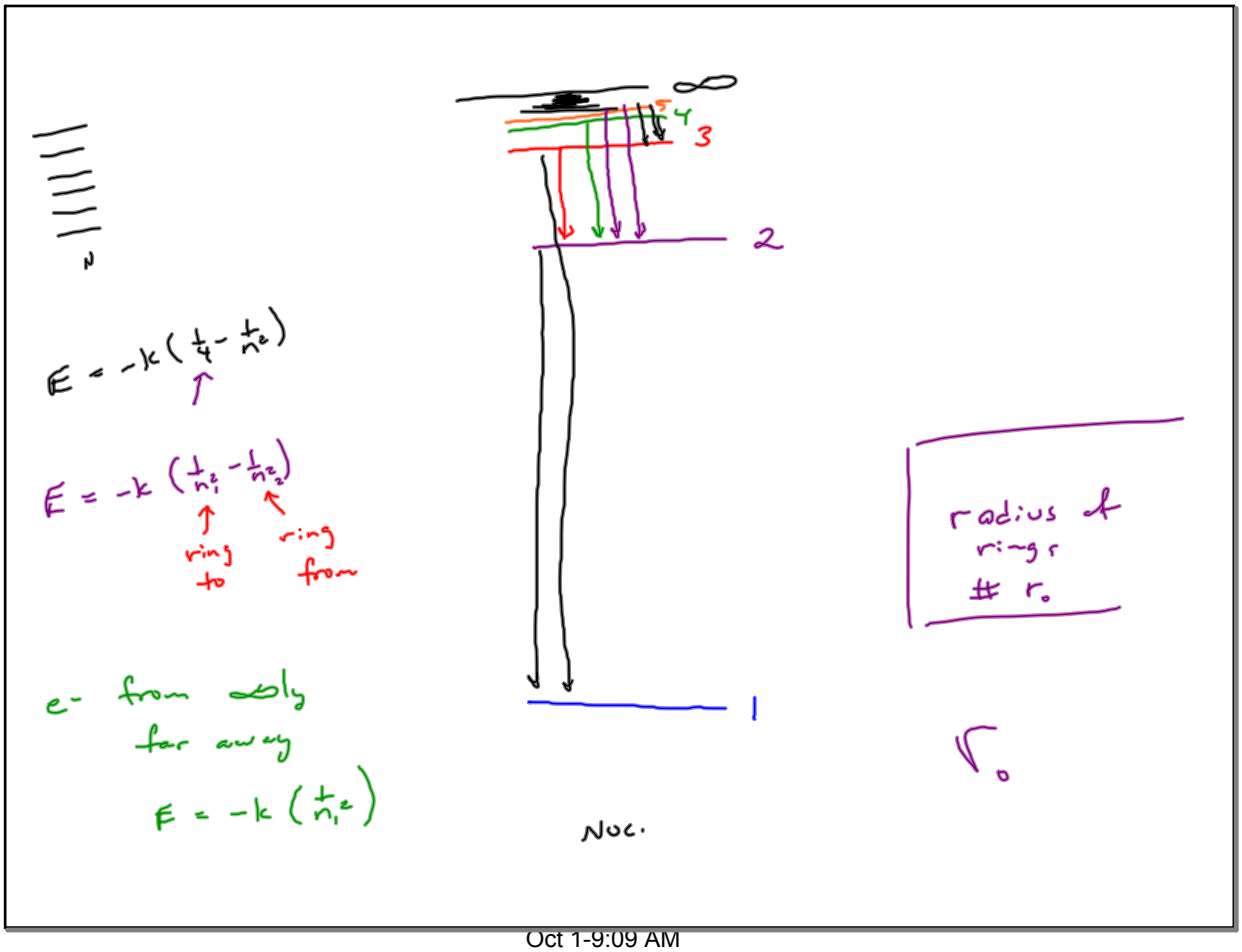


Sep 21-7:03 AM



Oct 1-9:00 AM


∞
5
4
3
2
1
Nuc.
E = -k(1/4 - 1/n²)
E = -k(1/n₁² - 1/n₂²)
ring to
ring from
e- from ∞ly far away
E = -k(1/n₁²)
radius of ring # r₀
r₀

Oct 1-9:09 AM